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to answer the
following: Mass of the
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Mass of the crucible
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Explain your answer.
The forward reaction is favored. The weaker acid and weaker base are produced in the forward reaction. HCO_3^- competes more strongly with H_2O to donate a proton, and OH^- competes more strongly with CO_3^{2-} to acquire a proton, causing the forward reaction to be favored.

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ACIDS AND BASES 121
Answers

**14 Acids and Bases -
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Sample Study Sheet

4.2: Abbreviated

Electron Configurations

To get a review of the most important topics in the chapter, fill in the blanks in the Key Ideas section. Work all of the selected problems at the end of the chapter, and check your answers with the

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solutions provided in this chapter of the study guide. Ask for help if you need it.

Chapter 4 Modern Atomic Theory - An Introduction to Chemistry

to answer the following: Mass of the empty crucible 4.00 g
Mass of the crucible plus hydrate sample 4.50 g
Mass of the system after heating 4.32 g
Mass of the

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system after a second heating 4.32 g 36% a.
Determine the mass percentage of water in the original sample.

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CHEMICAL FORMULAS
AND CHEMICAL
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7 Chemical Formulas and Chemical Compounds

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Introduction to
Chemistry Memorize
the following. (Be sure
to check with your
instructor to determine
how much you are
expected to know of
the following.) SI base
units (Table 1.1) The
table below contains

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the four base units that
you should know now.

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CHAPTER 10 REVIEW

States of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the

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crystal c covalent
molecular crystal (b) is
hard, brittle, and
nonconducting a
metallic crystal (c)
typically has the lowest
melting point of the
four

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